

# An effective fertilization program-

- · Promotes uniform growth
- Cost-effective

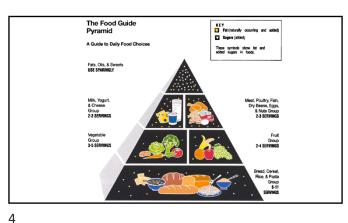
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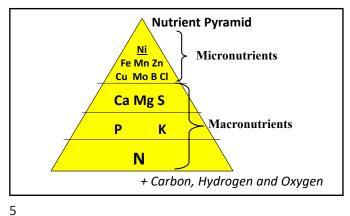
· As agronomically sound as possible

# **Topics**

- ➤ Nutrients: N, P, K **≻**Application Timing
- **≻**Choosing Products



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## Agronomic Law of Minimum

- "growth is proportional to the amount of the most limiting nutrient, whichever nutrient it may be"
- Macronutrients are not "more important" than micronutrients simply used in great quantity.

#### **Classification of Nutrients**

- Macro nutrients
  - 0.1 to 6.0% concentration by dry wt. of tissue
  - N 2-6%; P 0.25-1%; K 1-3%
  - Most common fertilizer needs
  - Structure of molecules
    - DNA, RNA, Chlorophyll

## **Classification of Nutrients**

Micronutrients

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- -<1.0 to 500 ppm concentrations
  - -ppm: 1 inch in 16 miles
- -Required in small quantities
- -Catalytic or regulatory roles
  - •Make reactions proceed/'go'

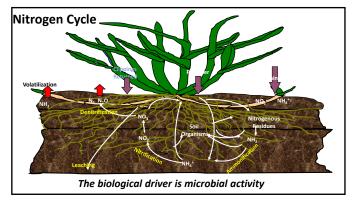
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# N Deficiency Symptoms

- Reduced growth rate, old leaves lose color
- Chlorosis (yellowing of leaves)
- Density loss



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Nitrogen Cycling

- Microbial activity influenced by:
  - Food source
  - Specific microbe
  - Soil pH
  - Soil aeration (oxygen or lack of)
  - Soil moisture (water)
  - Soil temperature

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Key Processes (N loss)

• Volatilization - loss of N as ammonia Urease

Urea +  $H_2O \longrightarrow CO_2 + NH_3$ 

- Factors:
  - · Urea level
  - · Urease enzyme present (thatch)
  - Temperature (slow < 50 F, increases to 77 F)

Phosphorus (P)

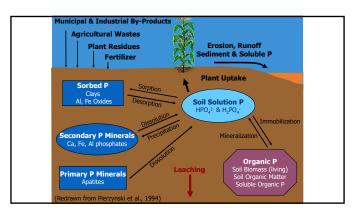
- Function
  - -Energy transformations (ATP, ADP)
  - -Constituent of genetic material

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# Plant Responses to P

- Root growth
- Shoot growth
  - P for energy transformations and structural components



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# P Deficiency Symptoms

- Reduced shoot growth
  - ATP/ADP involved in many metabolic reactions
- Dark green color
  - · Reduced leaf area, more chloroplasts/unit area,
- Red/purplish color
  - Anthocyanin accumulation







## **Phosphorus Applications**

- Apply according to soil test recommendations as part of annual program
- Critical for rooting initiation
- Apply at (re)seeding
- Apply after aeration
- Immobile so residual is high but may not be where the plant can use it (at the surface for seeding)

## Potassium (K)

- · Only nutrient that is not a constituent of any plant compound
- K is second behind only N in concentration in the plant
- What role does K serve in the plant?

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## Function

- Photosynthesis
   Carbohydrate and protein formation
   Enzymatic reactions
   Water relationships
   Enhances stress tolerance

# Plant Responses to K

- Root growth
  - deeper rooting & branching
- Stomatal control
  - K influences transpiration by regulating opening and closing of stoma
- Uptake of other nutrients
- Cell wall size and thickness
- Carbohydrate synthesis

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# K Deficiency Symptoms

- Leaves soft, drooping –reduced hardiness
- Interveinal yellowing of older leaves
- Reduced rooting
- Wilting
- Leaf margins brown
- Leaf tips roll
- Decreased tillering

#### **Potassium Applications**

- Sand rootzones apply as part of annual program
   Potassium leaches from sand rootzones
- Native profiles (medium, fine textured) soil test
  - Recommendations based on annual amounts
- General recommendation:
  - Sand rootzone: N:K ratio of 1:1
  - Native rootzone: N:K ratio of 1:0.5

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# Types of N Carriers

- 1. Fast/quick release water soluble
  - Salt based N carriers all types dissolve readily when adequate water is present forming either  ${\rm NO_3}^-$  or  ${\rm NH_4}^+$
  - Provide immediate nitrogen availability

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## Quickly Available N Fertilizers

#### **Advantages**

- less expensive
- quick response
- water-soluble
- not temperature dependent

#### Disadvantages

- peak and valley feeding
- higher burn potential
- higher labor costs

Examples: urea, ammonium sulfate, potassium nitrate

Quick Release N Carriers (Salt Types)

Carrier	Analysis	N Release
Ammonium nitrate (NH <sub>4</sub> NO <sub>3</sub> )	33-0-0	Water soluble
Ammonium sulfate ((NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> )	20-0-0	Water soluble
Potassium nitrate (KNO <sub>3</sub> )	13-0-44	Water soluble





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# Characteristics of Salt Types

- High salt index/high potential for burn
- Rapid response
- Apply at low rates, spoon feeding programs
- Inexpensive



## Slowly Available N Fertilizers

#### **Advantages**

- longer residual effect
- low burn potential
- less peak and valley feeding
- lowered N losses
- reduced labor

#### Disadvantages

- higher cost
- slower response
- most are temperature and/or moisture dependent

Examples: ureaformaldehyde, IBDU, methylene urea, sulfur- and resin-coated fertilizers, natural organics

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#### Hrea



- 46% N
- Soluble Synthetic Organic
- Nonionic, highly leachable
- Subject to volatilization
   Low acidity 1.8/kg N
- Low salt index 1.62

# UMAXX (47-0-0) and UFLEXX (46-0-0)

- Contains both nitrification (dicyandiamide) and urease (NBPT) inhibitors
- SuperN liquid mix of dicyandiamide and NBPT without urea



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#### Slow Release N Carriers

Synthetic organic N carriers - methylene urea based, provide a mixture of sources to provide both short- and long-term response



## Methylene Ureas

- Short chain molecules are water soluble
- Longer chain molecules are water insoluble
- Longer chain molecules N is released by microbial degradation

## Slow Release N Carriers

- 2. Natural organic N carriers
  - Derived from a variety of sources such as animal, plant, sewage, composts
  - N release is based on microbial degradation





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## Slow Release N Carriers

- 4. Coated N carriers
  - Several coating technologies including sulfur, resins, plastics, waxes, polymers
  - N release occurs with diffusion of water into the granule and diffusion of urea out of the granule into soil solution
  - Some temperature response



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Water Solul	ole vs. Wat Nitrogen	er Insoluble
	<b>WSN (Fast)</b>	WIN (Slow)
Cost	Low	High
Solubility	High	Low
Response	Quick	Slow
Residual Run off/	Short	Long
Leaching	High	Low
<b>Burn Potential</b>	High	Low
Appl. Frequency	High	Low

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N-SOURCE	SALT INDEX	RESIDUAL (WEEKS)
QUICK RELEASE		
Urea	1.62	4-6
Ammonium Nitrate	3.18	4-6
Ammonium Sulfate	3.25	4-6
SLOW RELEASE		
IBDU	0.20	6-8
Methylene Urea	0.86	6-8
Ureaformaldehyde	0.20	52+
Sulfur Coated Urea	0.70	Varies
Natural Organics	0.70	Varies

• Fertilizer blending takes advantage of the benefits of quickly and slowly available N sources
• Blending can make slowly available N sources more affordable

- Sources more affordab

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- Fertilizer selection depends economy, preference, labor and goals
- If slow-release fertilizers selected:
  - Apply early enough to provide available nitrogen at the expected period of vertical growth slowing or stoppage

# Timings for Carriers

- Soluble N sources: applied about the time growth ceases
- Sulfur coated ureas: 10 14 days prior
- Natural organics: 3 4 weeks prior
- IBDU: 4 6 weeks prior
- Mixed soluble and slow release:  $4-10\ \mbox{days}$  prior depending on % slow release and carrier

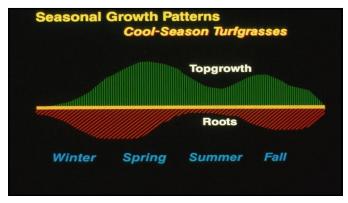
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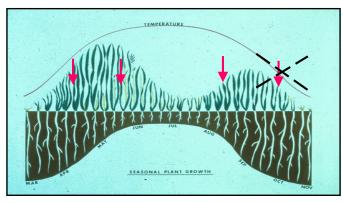
"Growth Potential" dictates application timing

 N carriers requiring soil microbial activity for N release must be applied 3 – 8 weeks prior to need in fall and spring in contrast in the summer where N availability may occur in 7 – 14 days.

Late Fall Fertility is no longer recommended

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# Why the change?

- Environmental
- Agronomic



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# Risks with Late Fall App's

• Runoff/Leaching

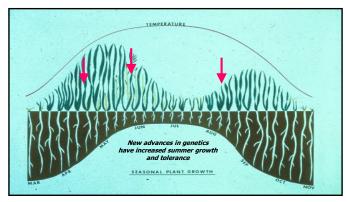


# Factors Affecting Leaching

- Soil temp. microbial nitrification
  - Frozen or impermeable?
- Soil moisture
- Drainage
- Turfgrass characteristics
- Soil texture (sand vs. clay)

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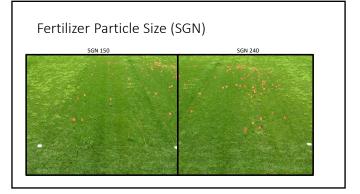




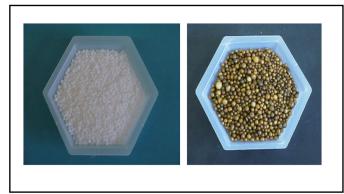
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# Factors Affecting Fertilizer "Value"

- Initial Cost
- Turf safety
- How long does it last?
- How easy is it to apply (spreadability)?
  - SGN
- Uniformity
- Incidental nutrients and other "stuff"



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# **Fertilization Tips**

- Fertilize to promote acceptable color, adequate growth, and rapid traffic recovery
- Use a combination of slow- and quick N sources
- Use iron if you want color without growth
- Let the grass tell you when it needs fertilization

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## Nitrogen Mineralization

- Release of significant plant available nitrogen occurs when soils are warm, moist and aerated.
- Nitrogen is mineralized from organic matter; older turf stands (with more soil organic matter) require less fertilizer (up to a 1 lb/yr).
- If clippings are returned, fertility can be reduced up to a 1 lb/yr.

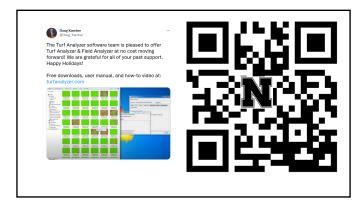
# Nitrogen Fertilization Considerations

- Don't fertilize turf that isn't actively growing.
- · Eliminate late fall fertility.
- Older lawns need less fertilizer and fewer applications annually than newer stands. Eliminate applications during high mineralization periods such as mid-summer or spring.
- Over-fertilized turf wastes money, can lead to excessive thatch accumulation, increased diseases and nitrate leaching to
- Different fertilizer sources have different release characteristics.

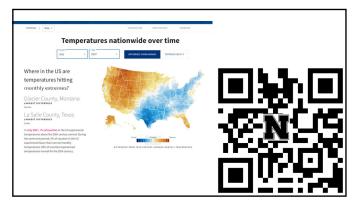
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# How Much?

- Depends on use and end goal
- Depending on stand age, cool season grasses in the central great plains need from 2-4 lbs/N/YR/M



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Thank you!